

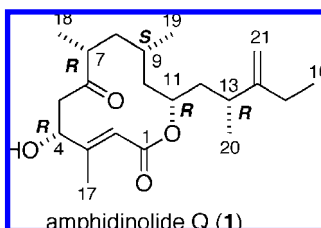
## Total Synthesis of Amphidinolide Q

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## ABSTRACT



Asymmetric synthesis of amphidinolide Q, a cytotoxic macrolide from the cultured dinoflagellate *Amphidinium* sp., has been accomplished with Julia coupling, Myers alkylation, and Yamaguchi lactonization. The absolute configuration of amphidinolide Q was confirmed to be **1** from comparison of the NMR data and  $[\alpha]_D$  values of synthetic and natural amphidinolide Q.

Amphidinolide Q (**1**) is a cytotoxic 12-membered macrolide having C1 branches at vicinal carbons (C-13 and C-14) and an  $\alpha,\beta$ -unsaturated ester moiety, isolated from the cultured dinoflagellate *Amphidinium* sp. (Y-5 strain).<sup>1</sup> Recently, we have proposed the stereoconfiguration of amphidinolide Q as **1** on the basis of extensive NMR experiments, molecular modeling, and chemical derivatization.<sup>2</sup> In this paper, we describe the first total synthesis of amphidinolide Q (**1**) and establish our proposed absolute stereochemistry.

As outlined retrosynthetically in Scheme 1, amphidinolide Q (**1**) could be obtained by Yamaguchi lactonization<sup>3</sup> of *seco*-acid **2**, which could be provided by aldol reaction of the C-1-C-5 segment (**3**) and the C-6-C-16 segment (**4**). Key aldehyde **4**, containing four stereogenic centers, could be derived from iodide **5** via Myers alkylation,<sup>4</sup> which is conceived to be obtained through Julia coupling<sup>5</sup> between sulfone **6** and aldehyde **7**.

The synthesis of iodide **5** is described in Scheme 2. Alcohol **8**<sup>6</sup> was transformed with  $(\text{PhS})_2\text{-Bu}_3\text{P}$ <sup>7</sup> into sulfide,

which was oxidized with *m*-chloroperoxybenzoic acid to sulfone **6**. Alcohol **9**<sup>8</sup> was oxidized with Dess–Martin periodinane<sup>9</sup> to the corresponding aldehyde **7**, which was then subjected to Julia coupling<sup>5</sup> with **6** to afford hydroxy sulfone. Ketone **10** was obtained following oxidation and reductive removal of the sulfone group.<sup>10</sup> Reduction of ketone **10** with  $\text{NaBH}_4$  gave diols **11** and **12** (37% and 32%, respectively). Selective protection of the primary hydroxy group in **11** provided pivaloate ester **13**, the secondary hydroxy group of which was treated with MOMCl and *i*-Pr<sub>2</sub>NEt to afford MOM ether **14**. After removal of the pivaloyl group in **14**, the corresponding alcohol was oxidized to an aldehyde, then treated with EtMgBr and oxidized to yield ketone **15**. Wittig olefination was followed by deprotection and iodination to afford iodide **5**.

The absolute configuration at C-11 in **13** was elucidated by a modified Mosher's method.<sup>11</sup> Treatment of **13** with (*R*)-(-)- and (*S*)-(+)-2-methoxy-2-trifluoro-2-phenylacetyl chloride (MTPACl) provided the (*S*)- and (*R*)-MTPA esters (**13a**

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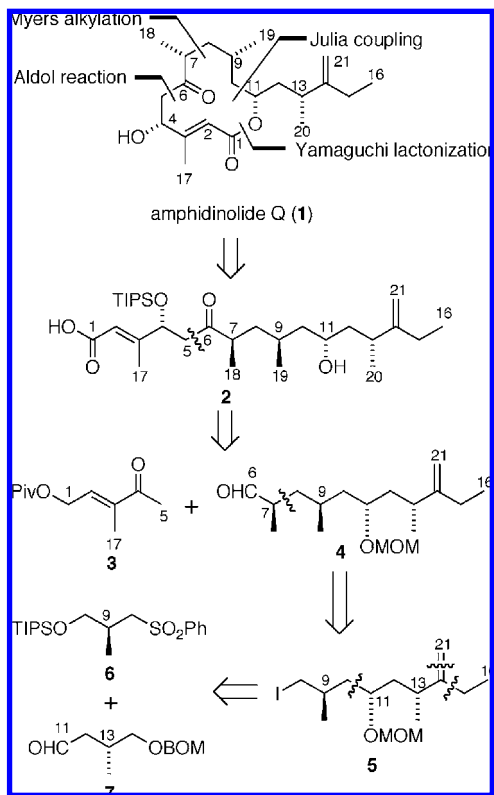
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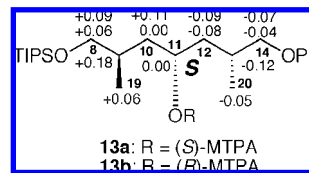
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**Scheme 1.** Retrosynthetic Analysis of Amphidinolide Q (**1**)



and **13b**, respectively) of **13**.  $\Delta\delta$  values ( $\Delta\delta = \delta_S - \delta_R$ ) obtained from  $^1\text{H}$  NMR data of **13a** and **13b** are shown in

Figure 1. The  $\Delta\delta$  values for H<sub>2</sub>-8, H-9, H<sub>2</sub>-10, and H<sub>3</sub>-19



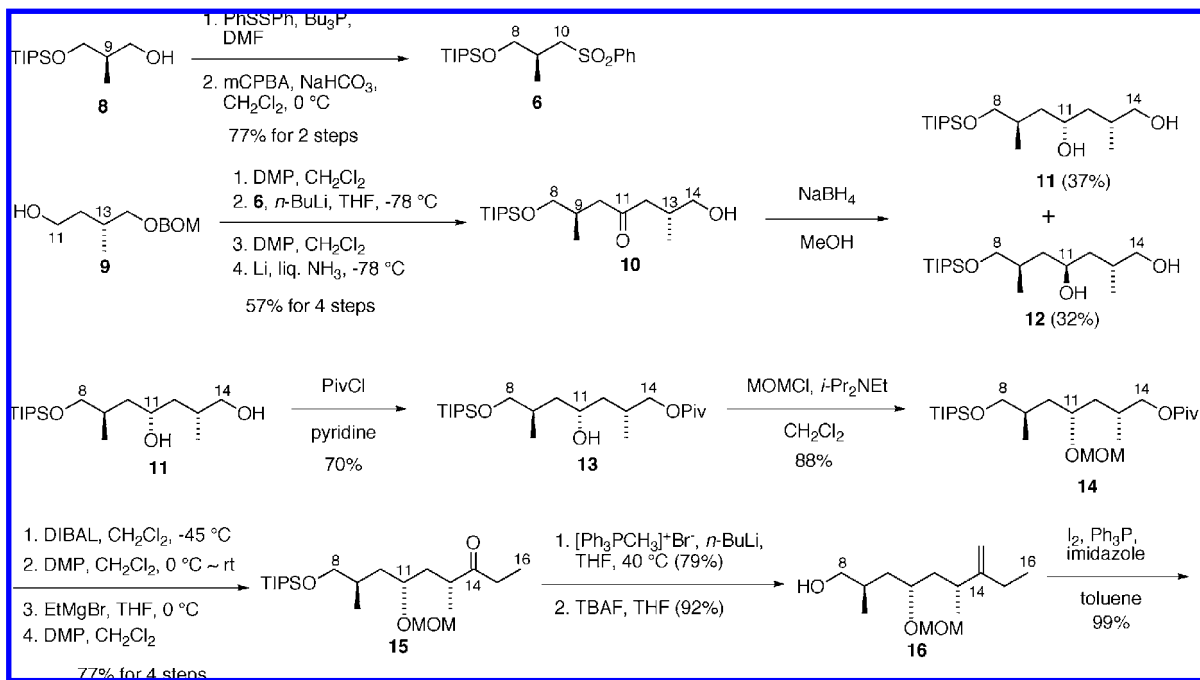
**Figure 1.**  $\Delta\delta$  values [ $\Delta\delta$  (in ppm) =  $\delta_S - \delta_R$ ] obtained for (S)- and (R)-MTPA esters at C-11 (**13a** and **13b**, respectively) of alcohol **13**.

were positive, while negative  $\Delta\delta$  values are observed for H<sub>2</sub>-12, H-13, H<sub>2</sub>-14, and H<sub>3</sub>-20. These results indicated that the absolute configuration at C-11 was *S*.

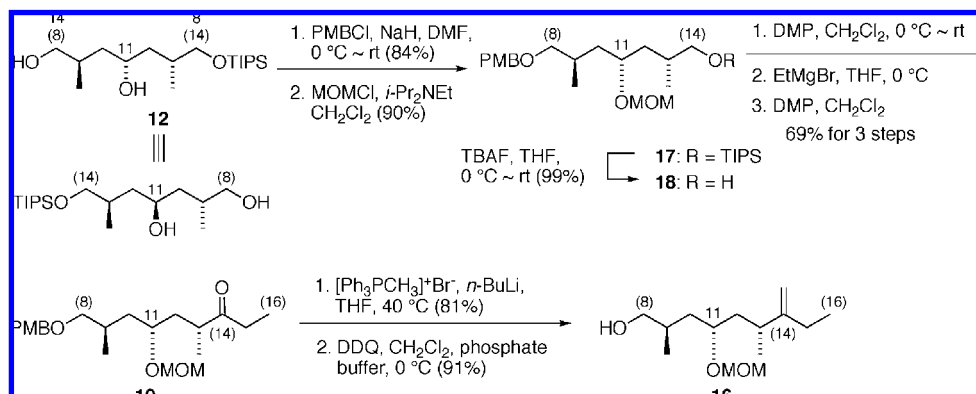
As shown in Scheme 3, alcohol **12** was also converted into **16**. Selective hydroxy group protections furnished **17**. After removal of TIPS ether in **17**, alcohol **18** was oxidized to the corresponding aldehyde and then treatment with EtMgBr afforded alcohol, which was oxidized with Dess–Martin periodinane to ketone **19**. Wittig reaction to install an exomethylene followed by treatment of DDQ yielded alcohol **16**, which was transformed into iodide **5** as described in Scheme 2.

Alcohol **20**<sup>12</sup> was protected as pivaloate ester to afford the C-1-C-5 segment (**3**) (Scheme 4). Myers alkylation<sup>4</sup> was used to install the C-7 stereocenter essentially as a single diastereomer. Reductive cleavage of the auxiliary by using LDA-BH<sub>3</sub>·NH<sub>3</sub> complex provided alcohol **22**. Oxidation of **22** with Dess–Martin periodinane and then aldol reaction of

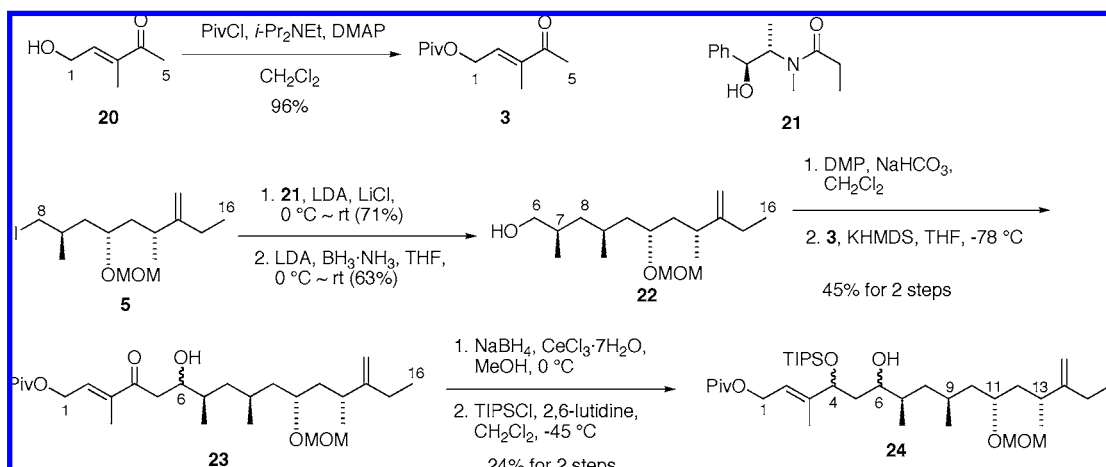
**Scheme 2.** Synthesis of the C-8-C-16 Segment (**5**) from **8**



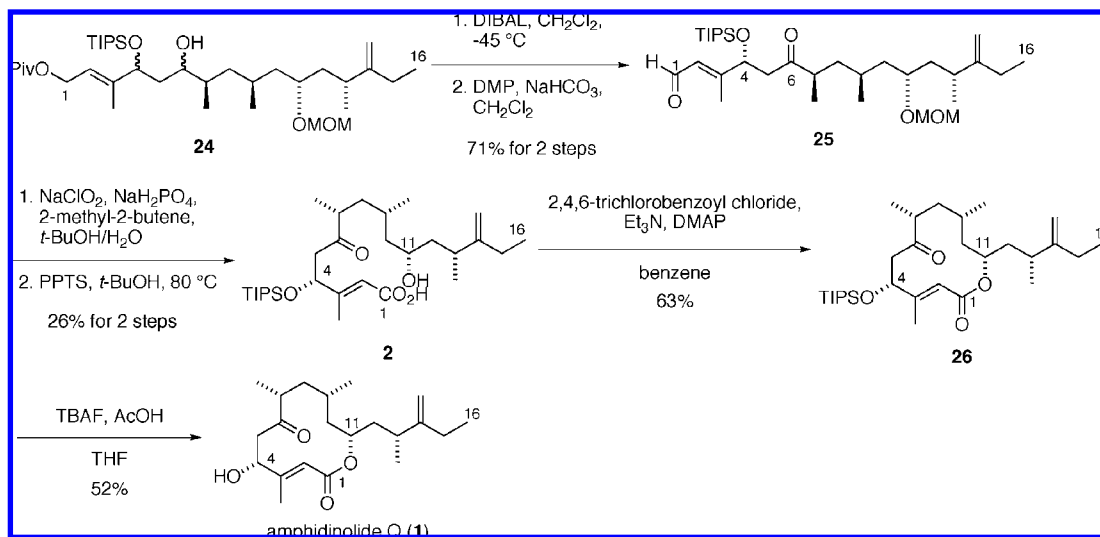
**Scheme 3.** Synthesis of the C-8-C-16 Segment (**16**) from **12**



**Scheme 4.** Synthesis of the C-1-C-16 Segment (**24**) from **5**



**Scheme 5.** Synthesis of Amphidinolide Q (**1**) from **24**



the corresponding aldehyde and **3** with KHMDS afforded  $\beta$ -hydroxy ketone **23**. Reduction of **23** with NaBH<sub>4</sub> and CeCl<sub>3</sub>·7H<sub>2</sub>O followed by selective protection of the allylic hydroxy group yielded TIPS ether **24** as a diastomeric mixture.

Removal of the pivaloyl group in **24** with DIBAL followed by oxidation with Dess–Martin periodinane provided aldehyde **25** as a single diastereomer after silica gel column separation (Scheme 5). After aldehyde **25** was oxidized under Pinnick oxidation conditions<sup>13</sup> to carboxylic acid, the MOM group was removed with PPTS to afford *seco*-acid **2**. The *seco*-acid (**2**) was then subjected to macrolactonization by using the Yamaguchi procedure<sup>3</sup> to provide macrolactone

**26**. Finally, removal of the TIPS group in **26** with TBAF and AcOH furnished amphidinolide Q (**1**). The absolute configuration at C-4 in **1** was confirmed by a modified Mosher's method as in the previous report.<sup>2</sup> Synthetic amphidinolide Q (**1**) was identical with natural amphidinolide Q (<sup>1</sup>H and <sup>13</sup>C NMR, IR, UV, MS, and optical rotation),<sup>1,2</sup> thus allowing confident assignment of the absolute configurations and validating our earlier proposal.<sup>2</sup>

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**Supporting Information Available:** Experimental details and spectroscopic data. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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